PRESS RELEASE

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(Tokyo, January 8) Scientists at Tokyo Institute of Technology (Tokyo Tech) and the National Taiwan University of Science and Technology (Taiwan Tech) develop a cheap and efficient copper-based catalyst that can be used to convert glycerol, one of the main by-products of the biodiesel industry, into a valuable compound called dihydroxyacetone. In addition, this same process produces hydrogen molecules from water, and these could be used as a clean type of fuel, further highlighting the impact of this research in terms of energy sustainability.

Although governments, academia, and organizations all around the world have been emphasizing the crisis concerning the use of fossil fuels for many years, their demand has constantly been on the increase. Now that supply is seriously dwindling, researchers have fervently focused on finding alternative fuels that are cleaner and with the potential for sustainable production.

Hydrogen (H₂) is a very attractive candidate as a replacement of fossil fuels because it can be produced from water (H₂O) through hydrolysis, the splitting of water molecules. Another sustainable route is the synthesis of biodiesels, which are made using vegetable oils through a transformation process known as transesterification. However, biodiesel synthesis produces excessive amounts of glycerol (C₃H₈O₃); it is estimated than the biodiesel industry in Europe alone produces a surplus of 1.4 million tons of glycerol, which cannot be sold to other industries. If glycerol could be used as a raw material to obtain more valuable chemicals, this would make the biodiesel industry more profitable, thus encouraging governments and companies to switch away from fossil fuels.

Fortunately, researchers from Tokyo Tech and Taiwan Tech recently found an efficient way to put this surplus glycerol to good use. While the electrochemical conversion of glycerol to other more valuable organic compounds, such as dihydroxyacetone (DHA), has been studied for years, existing approaches require the use of precious metals, namely platinum, gold, and silver. Because the use of these metals represents 95% of the overall cost of glycerol to DHA conversion, this research team focused on finding an affordable alternative.

In their study, they found that copper oxide (CuO), a cheap and abundant material, could be used as a catalyst to selectively convert glycerol into DHA even at mild reaction conditions. For this to happen, the pH (concentration of free hydrogen ions) in the solution of the electrochemical cell has to be at a specific value. Through various microscopy techniques, the researchers analyzed the crystalline structure and composition of the CuO catalyst and tailored them to make it stable while also carefully inspecting the possible conversion pathways for glycerol in their system according to the solution's pH. This allowed them to find appropriate reaction conditions that favored the production of DHA. "We have not only discovered a new, earth-abundant catalyst for high-selectivity DHA conversion, but also demonstrate the possibility of giving new valuable life to a waste product of the biodiesel industry," remarks Prof. Tomohiro Hayashi, lead researcher from Tokyo Tech.

What's more, the electrochemical system proposed in this study not only produced DHA from glycerol on one end, but also H₂ on the other through water splitting. This means that this approach could be used to address two current problems simultaneously. "Both the biodiesel and the hydrogen generation industries could benefit from our system, leading to a more sustainable world," explains Prof. Hayashi. A diagram of sustainable energy cycles including both the biodiesel and hydrogen industries is shown in Figure 1. In conclusion, it is crucial that we keep on trying to address the problem of sustainability in our use of fuels, and studies such as this one take us one step closer to a greener future.

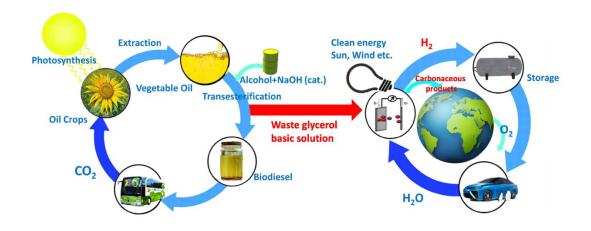


Figure 1. Sustainable biodiesel and hydrogen energy cycles

One of the main waste by-products of the biodiesel industry, glycerol, can be used as a raw material for the generation of valuable dihydroxyacetone and hydrogen, the latter of which can be used as 100% clean fuel.

Reference

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Title of original paper: Journal: DOI: Affiliations:	Selective electro-oxidation of glycerol to dihydroxyacetone by a non-precious electrocayalyst—CuO Applied Catalysis B: Environmental 10.1016/j.apcatb.2019.118543 ¹ Department of Chemical Engineering, National Taiwan University of Science and
Annations.	Technology ² Department of Materials and Engineering, School of Chemical Technology, Tokyo Institute of Technology ³ JST-PRESTO

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